

–HOAc); 1 A + 1 177 (23); 169 (41) (271 – ketene and HOAc); 109 (35) (169 – HOAc).

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## A XANTHONE-*O*-GLYCOSIDE FROM *ASPLENIUM ADIANTUM-NIGRUM*

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**Key Word Index**—*Asplenium adiantum-nigrum*; Aspleniaceae; 3,7,8-trihydroxyxanthone-1-*O*- $\beta$ -laminaribioside.

**Abstract**—A new xanthone-*O*-glycoside isolated from the fern *Asplenium adiantum-nigrum* has been identified as 3,7,8-trihydroxyxanthone-1-*O*- $\beta$ -laminaribioside by chemical and spectroscopic methods.

Recently a new hydroxycinnamic acid–sugar derivative, 1-caFFEyllaminaribiose has been characterized from *Asplenium adiantum-nigrum* L. [1]. The present paper describes the identification of a new xanthone-*O*-glycoside, which has the same dissaccharide present, from the same fern.

The xanthone (colour reactions: orange–brown to yellow in UV + NH<sub>3</sub>) was isolated by prep. PC of an ethanolic extract of the fronds of *Asplenium adiantum-nigrum*. The UV spectral data:  $\lambda_{\text{max}}^{\text{MeOH}}$  240, 260, 315 and 364 nm; + NaOAc 260, 360 nm (increase in intensity); + AlCl<sub>3</sub> 245, 264, 349 (sh), 405 nm; + AlCl<sub>3</sub>/HCl 243, 262, 340 (sh), 405 nm; + NaOAc/H<sub>3</sub>BO<sub>3</sub> 242, 258, 355 (sh), 418 nm, are consistent with a xanthone skeleton. The presence of a 3-hydroxyl group and an *ortho*-dihydroxyl group in the 7,8-position is indicated by the shifts with NaOAc, AlCl<sub>3</sub>, AlCl<sub>3</sub>/HCl and H<sub>3</sub>BO<sub>3</sub>/NaOAc [2]. Both total acid hydrolysis and treatment with  $\beta$ -glucosidase gave D-glucose and an aglycone identified as 1,3,7,8-tetrahydroxyxanthone (norswertianin). This xanthone was first isolated in the free state from *Swertia japonica* [3] but has since been reported from *Gentiana bavarica* L. [4]. Norswertianin has been found as the 1-*O*-glucoside and 1-*O*-primeveroside in some *Swertia* and *Gentiana* species [5]. Controlled acid hydrolysis gave D-glucose and a

disaccharide which was identical with an authentic sample of laminaribiose (3-*O*- $\beta$ -glucosyl-D-glucose) synthesized according to Bächli and Percival [6]. Thus the isolated compound must be 3,7,8-trihydroxyxanthone-1-*O*- $\beta$ -laminaribioside, a new natural product. The structure of this substance was confirmed by methylation followed by acid hydrolysis to give 2,4,6-tri-*O*-methyl-D-glucose, 2,3,4,6-tetra-*O*-methyl-D-glucose and a partially methylated aglycone identified as 1-hydroxy-3,7,8-trimethoxyxanthone (decussatin) [7–9]. Xanthenes have only been reported twice in ferns, in *Asplenium montanum* [10] and *Athyrium mesosorum* [11, 12]. However, the presence of a xanthone-*O*-glycoside in ferns is here reported for the first time. The absence of xanthenes in *Asplenium adiantum-nigrum* L. collected in Asturias, Spain [10] was not confirmed in the present study. The chemical differences in plants from this locality may be due to phytogeographical factors. Since xanthenes have been found previously only in *Asplenium montanum* among the several species of *Asplenium* surveyed [10], the isolation of a xanthone from *A. adiantum-nigrum* L. may suggest a relationship to *A. montanum*. Laminaribiose is here reported for the first time in association with xanthenes. The other disaccharides found to date in xanthone-*O*-glycosides [5] are rutinose and primeverose.

## EXPERIMENTAL

**Plant material.** Fronds of *Asplenium adiantum-nigrum* L. were collected on Mount Etna, Sicily.

**Isolation procedure.** Fresh fronds of *Asplenium adiantum-nigrum* were homogenized and extrd  $3 \times$  with hot EtOH. The combined extracts were filtered, concd to small vol. *in vacuo* and re-filtered. The xanthone was isolated by successive prep. PC in BAW, 5% HOAc and BEW.  $R_f$  data are: BAW 0.45, 5% HOAc 0.25, BEW 0.17. Total acid hydrolysis was carried out with 2 N HCl (1 hr at 100° under  $N_2$ ); controlled acid hydrolysis was carried out with 10% aq. HOAc (2.5 hr under reflux). Treatment with  $\beta$ -glucosidase was carried out in citrate-phosphate buffer, pH 4.5, at 37° for 20 hr. 1,3,7,8-Tetrahydroxyxanthone was identified by UV spectral analysis with shift reagents [2,4], MS and comparison with an authentic sample (TLC on Si gel: 3 solvents); D-glucose and laminaribiose were identified by Co-PC (4 solvents), TLC on Si gel (*n*-BuOH-HOAc-Et<sub>2</sub>O-H<sub>2</sub>O, 9:6:3:1) and GLC of their trimethylsilyl derivatives [13].

**Methylation of xanthone.** The xanthone was methylated ( $Me_2SO_4$ - $K_2CO_3$ - $Me_2CO$ ) and hydrolysed with 0.3 N HCl (4 hr under reflux). The partially methylated aglycone was identified as 1-hydroxy-3,7,8-trimethoxyxanthone by UV spectral analysis with shift reagents [2,4], MS and comparison with an authentic sample (TLC on Si gel; 3 solvents); methylated sugars were identified by PC according to ref. [14] and TLC on Si gel (EtOAc- $CHCl_3$ , 1:1).

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## AN UNSYMMETRICAL DIARYLHEPTANOID FROM *CURCUMA LONGA*

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**Key Word Index**—*Curcuma longa*; Zingiberaceae; turmeric; diarylheptanoid; dihydrocurcumin; 1,7-bis-(4-hydroxy-3-methoxyphenyl)-hept-1-en-3,5-dione.

Earlier work on the pigments of turmeric (*Curcuma longa* L.) revealed the presence of three major phenolic diarylheptanoids, namely curcumin (1), feruloyl-(4-hydroxycinnamoyl)methane (2) and bis-(4-hydroxycinnamoyl)methane (3) [1,2]. Biosynthesis and metabolism of diarylheptanoids have recently attracted attention [2,3]. In the course of our work on the metabolism of curcumin [4], we have isolated a new diarylheptanoid from the benzene extract of *Curcuma longa* rhizomes.

This compound,  $C_{21}H_{22}O_6$  ( $M^+$  370), had a UV

maximum at 375 nm and its IR spectrum showed bands at 3400 (OH), 1630 ( $COCH_2CO$ ), 1600 and  $1510\text{ cm}^{-1}$  ( $C=C$  and aromatic). It gave the rubrocurcumin reaction [5] with boric acid and oxalic acid (with visible max shifting to 420 nm), a characteristic reaction of  $\beta$ -diketones. The 270 MHz  $^1H$  NMR spectrum in  $DMSO-d_6$  showed signals for two phenolic hydroxyl protons at  $\delta$ 9.65 and 8.76 (disappearing on exchange with  $D_2O$ ), two singlets at 3.74 (3 H) and 3.83 (3 H) (OMe), two doublets at 7.48 and 6.64 ( $J = 14\text{ Hz}$ ) for two *trans*-related olefinic protons, clusters of aromatic proton signals centred around 6.65,